STUDIES OF THE POLYSACCHARIDES OF SUGAR CANE (Saccharum officinarum): STRUCTURAL FEATURES OF THE WATER-INSOLUBLE D-XYLANS*

FRANCISCO SAAVEDRA,

Instituto de Investigaciones de la Celulosa del Bagazo, Cuba-9 ICIDCA, P.O. Box 4026 Ciudad Habana (Cuba)

ŠTEFAN KARÁCSONYI, AND JURAJ ALFÖLDI

Institute of Chemistry, Slovak Academy of Sciences, 842 38 Bratislava (Czechoslovakia)
(Received August 21st, 1987; accepted for publication, January 30th 1988)

ABSTRACT

Polysaccharides extracted from the cell-wall material of Saccharum officinarum were fractionated into pectic polysaccharides, hemicellulose, and cellulose components. Structural features of the alkali- and methyl sulfoxide-soluble, water-insoluble arabinoglucuronoxylan, arabinoxylan, and O-acetylarabinoglucuronoxylan were established by methylation analysis and ¹³C-n.m.r. spectroscopy. The results show that the D-xylans exhibit heterogeneity and have structural features similar to those found in non-endospermic tissues of Gramineae.

INTRODUCTION

Sugar cane (Saccharum officinarum) is a widely distributed member of the Gramineae and an important source of foodstuff. Moreover, its fibrous residue, bagasse, is a raw material for the production of cellulose, paper, boards and 2-fural-dehyde, but there have been few studies of its content of polysaccharides¹⁻³. We now report on the water-insoluble polysaccharides.

RESULTS AND DISCUSSION

Extractive-free cell walls were depectinated with hot water and ammonium oxalate and then delignified⁴. Treatment of the resulting holocellulose (81.7% of the wall) with alkaline solutions solubilised the hemicelluloses, which were fractionated into water-soluble and water-insoluble materials. The proportions of the polysaccharide fractions and their chemical composition are given in Table I.

The water-insoluble polysaccharide fractions C-1 and D-1, which constitute

^{*}Presented, in part, at the 4th European Carbohydrate Symposium, Darmstadt, F.R.G., July 12-17, 1987.

TABLEI

EXTRACTION OF POLYSACCHARIDES FROM Saccharum officinarum (ELL-WALL MATERIAL (CWM)

	Extractant	Yield	$[\alpha]_{D^o}$	Uronic	Molar n	Molar ratios of monosaccharides	nosacchar	ides			
		(%)	ε	acid (%)	Gal	Glc	Man	Ara	Xyl	Rha	Fuc
CWMc				1.06	9.0	18.1	9.0	1.4	10.0	0.1	0.1
¥	Water	1.31	+21	1.20	14.1	18.4	12.1	5.5	10.0	9.81	12.3
В	aq. 0.5% ammonium oxalate	1.01	+61	1.86	21.5	113.4	20.0	9.9	10.0	24.5	7.2
$C-1^d$	aq. 15% potassium hydroxide	30.6	-136	1.08	0.1	1.3	0.1	9.0	10.0		
C-2¢	•	6.1	9/-	1.80	0.5	1.3	0.1	2.3	10.0	0.1	
$D-1^d$	aq. 17.5% sodium hydroxide +	Ξ:	-111	5.30	Ħ	1.0	0.3	0.7	10.0		
D-2¢	4% of boric acid	0.4	-45	2.76	1.1	6.3	4.5	3.1	10.0	0.2	
П	Residuc	39.1				100.4		9.0	1.0		

"Based on CWM. hAqueous 4% potassium hydroxide, 'Cellulose, 44.9; lignin (Klason), 20.5; acetyl, 0.3%, "Water-insoluble material, 'Water-soluble material.

TABLE II	
O-ACETYLATED POLYSACCHARIDES FROM Saccharum officinarum	CWM

Fraction	Extractant	Yield ^a (%)	$[\alpha]_{\mathrm{D}}^{b}$	Acetyl	Molar	ratios of n	nonosacch	arides	
		1,	1 /	()	Gal	Glc	Man	Xyl	Ara
P-1	Me ₃ SO	3.3	-100	6.75		0.1		10.0	0.7
P-2	water	1.1	-79	3.87	0.4	0.5	0.1	10.0	1.5

[&]quot;Based on CWM. bAqueous 4% potassium hydroxide.

TABLE III

CHEMICAL COMPOSITION OF NEUTRAL OLIGOSACCHARIDES ISOLATED FROM ARABINOGLUCURONOXYLAN (AGX)

Saccharide	Yield (mg)	D.p.	$[\alpha]_{\mathrm{D}}^{a}$	R_{xyl}^b	Methylation an	alysis (mol %)	
	(<i>mg</i>)		· / /		2,3,4-Me ₃ -Xyl	2,3-Me ₂ -Xyl	2,4-Me ₂ -Xyl
1 ^c	20.2	2	-26	0.70	44.6	49.6	5.8
2	29.0	3	-47	0.44	34.3	65.7	
3	9.5	4	59	0.32	19.9	80.1	
4	7.0	5	-70	0.15	nd^d	nd	nd
5	9.2	6	-78	0.07	nd	nd	nd

[&]quot;Water. ^bP.c. (solvent D). 'This compound also contained small amounts (<0.15 mol%) of ($1\rightarrow3$)-linked xylobiose. ^dNot determined.

31.4% of the cell wall, were fractionated repeatedly by treatment with Fehling's solution⁵. This process yielded an arabinoglucuronoxylan AGX, $[\alpha]_D$ -143° (aqueous KOH), which consisted of xylose and arabinose in the molar proportions 10.0:0.6 together with 1% of uronic acid. A neutral arabinoxylan AX was also obtained with $[\alpha]_D$ -132° and a molar ratio of xylose and arabinose of 10.0:0.5.

Treatment of the holocellulose with methyl sulfoxide⁶ and subsequently with water gave the partially acetylated hemicellulose fractions P-1, $[\alpha]_D - 100^\circ$ (xylose-arabinose 10:0.7; OAc, 6.8%), and P-2, $[\alpha]_D - 41^\circ$ (xylose-arabinose 10:1.5; OAc, 3.9%) (see Table II for yields and analytical data).

Partial hydrolysis of AGX with acid gave acidic and neutral fractions. The neutral fraction contained the saccharides 1–5 (Tables III and IV), which were isolated using Sephadex G-10 and by p.c. and identified on the basis of methylation analysis, mass spectrometry, and 13 C-n.m.r. spectroscopy as 4-O- β -D-xylopyranosyl-D-xylose (1) and the corresponding tri- (2), tetra- (3), penta- (4), and hexa-saccharide (5). Compounds 1–5 were also identified in the partial hydrolysate of AX.

The acidic fraction contained the aldouronic acids 6-9 (Table V) which were characterised by ¹³C-n.m.r. spectroscopy as 2-O-(4-O-methyl-α-D-glucopyranosyl-

TABLE IV

¹³C-N.M.R. DATA FOR NEUTRAL OLIGOSACCHARIDES ISOLATED FROM ARABINOGLUCURONOXYLAN (AGX)

Saccharidea	Residue	Chemical	Chemical shifts (8, p.p.m.)	т.)		
		<i>I-D</i>	C-2	C-3	C-4	C-5
1 β·D·Xy[p·(1→4)·D·Xy]p	А	α 93.2	72.6	72.1	77.8	60.1
B A		β 97.7	75.1	75.2	72.7	64.2
	В	103.0	74.0	8.92	70.4	66.4
2 β -D-Xylp-(1 \rightarrow 4)- β -D-Xylp-(1 \rightarrow 4)-10-Xylp	4	α 93.2	72.6	72.1	77.8	60.1
C B A		β 97.7	75.1	75.2	77.6	64.2
	В	102.9	73.9	74.9	77.6	64.2
	C	103.0	74.0	8.92	70.4	66.4
$3\beta_{\text{-D-Xyl}p}$ - $(1\rightarrow 4)\cdot\beta_{\text{-D-Xyl}p}$ - $(1\rightarrow 4)\cdot\beta_{\text{-D-Xyl}p}$	Ą	α 93.2	72.6	72.1	77.8	60.1
C B B A		B 97.7	75.1	75.2	77.6	64.2
	В	102.8	73.9	74.9	9.77	64.2
	၁	103.0	74.0	76.8	70.4	66.4

^aCompounds 4 and 5 showed the same ¹³C-n.m.r. spectra as 3.

TABLE V

ACIDIC OLIGOSACCHARIDES FROM ARABINOGLUCURONOXYLAN (AGX)

Saccharide	Yield	D.p.	$[\alpha]_{D^a}$	\mathbb{R}_{GlcA}^{b}	Residue	Chemical shifts (6, p.p.m.) ^d	shifts (6, p	.p.m.) ^d				The state of the s
The state of the s	(8111)	Tagger and the second s				C-1	C-2	C-3	C-4	C:S	Q-Q	MeO-4
9	18.7	7	+86.5	0.67	t-Xylp	a 91.0	7.77	72.6	70.6	62.0		
					GlcA	98.7	72.6	73.4	83.6	70.6	172.3	61.1
7	24.0	ю	+57	0.34	t-Xylp	α 93.1	72.5	72.5	77.6	59.8		
					s-Xylp	р 97.7 102.7	78.0	75.6	6.// 70.6	64.0 66.1		
					GlcA	9.86	72.5	73.5	83.6	9.07	172.3	61.1
∞	22.6	4	+25	0.15	t-Xylp	α 93.1	72.5	72.5	77.6	59.8		
					i-Xylp	β 97.7 102.8	74.9	73.9	77.6	64.0 0.4.0		
					s-Xylp GlcA	102.8	78.4	77.6	70.6	66.1	177.3	61.1
6	8.01	5	+0.6	0.05		Š	ì	1				7.17
	-	-	-				PROPERTY					The same of the sa

«Water. bP.c. (solvent E). 4, terminal; s, substituted; i, internal. Downfield from the signal for Me4Si, measured against internal methanol (850.15).

IADLE VI						
METHYLATED SUC	GARS FROM	HYDROLYSATES	OF THE	METHYLATED	XYLANS AND	O-ACETYLARABINO-
GLUCURONOXYLA	.N					

Sugar	\mathbf{T}^{b}	Molar ra	tios		Linkage indicated
(as alditol acetate)		AGX	AX	P-1	
2,3,5-Me ₃ -Araf ^u	0.68	0.93	0.53	1.60	L-Ara <i>f-</i> (1→
2,3,4-Me ₃ -Xylp	0.83	1.00	1.00	1.00	$D-Xylp-(1\rightarrow$
$2,3-Me_2-XyIp$	1.16	13.50	9.52	20.00	$\rightarrow 4$ -D-Xylp-(1 \rightarrow
3.4-Me ₂ -Xylp	1.03	0.64			$\rightarrow 2-0-Xylp-(1\rightarrow$
3-Me-Xylp	1.43	0.20	0.18	0.30	$\rightarrow 4.2$ -D-Xyl p -(1 \rightarrow
2-Me-Xylp	1.43	1.83	1.59	2.70	$\rightarrow 4.3$ -D-Xyl p -(1 \rightarrow

[&]quot;2,3,5-Me₃-Araf = 1,4-di-O-acetyl-2,3,5-tri-O-methylarabinitol, etc. ^bRetention times of the corresponding alditol acetates with respect to that of 1,5-di-O-acetyl-2,3,4,6-tetra-O-methyl-D-glucitol on column B (see Experimental).

uronic acid)-D-xylose (6), and (1 \rightarrow 4)-linked D-xylobiose (7), D-xylotriose (8), and D-xylotetraose (9) with 4-O-methyl- α -D-glucuronic acid linked to O-2 of the non-reducing terminal unit. The aldobiouronic acid 6 has been isolated from various Gramineae xylans, e.g., corn cob⁷, cocksfoot grass⁸, and jute fibre⁹; in addition to 6, 2-O-(α -D-glucopyranosyluronic acid)-D-xylose has been found in hydrolysates of xylans from wheat straw¹⁰ and oat straw¹¹.

Methylation of AGX by the Hakomori procedure¹² and twice by the Purdie method¹³, followed by g.l.c.-m.s.¹⁴ of the derived alditol acetates, revealed derivatives of 2,3,5-tri-O-methylarabinose, 2,3,4-tri-O-methylarylose, 2,3-di-O-methylarylose, and 2(3)-O-methylarylose (see Table VI). These results suggested that AGX contained a backbone of (1 \rightarrow 4)-linked xylopyranosyl residues with more than two of every seventeen residues substituted by terminal arabinosyl and xylosyl groups mainly at O-3. Small amounts of 3,4-di-O-methylarylose (Table VI, column 1) indicated the presence of some (1 \rightarrow 2) linkages. The average AGX molecule contained 168 glycosyl residues, thus suggesting the presence of 18 branch points.

Linkage analysis of AX and P-I revealed structural features the same as those reported for AGX (Table VI), but with different proportions of constituent sugars and pattern of substitution. The average AX and P-I molecules contained 160 and 142 glycosyl residues, respectively, and thus it is concluded that they have 20 and 16.6 branching points, respectively.

The 13 C-n.m.r. spectra of the partially acetylated and deacetylated P-I and AGX (see Table VII) differ by the presence, in the former, of signals characteristic of acetyl groups (20.1, 20.8, 169.2, and 168.9 p.p.m.). The precise assignments were effected on the basis of the literature data $^{15-18}$ and the spectral characteristics of 1-9. The positions of the O-acetyl groups were confirmed using methyl 3- and 2-O-acetyl- β -D-xylopyranosides as model compounds to represent the (1 \rightarrow 4)-linked β -D-xylopyranosyl residues in the native acetylated xylan macromolecule 19 .

TABLE VII

13C-n.m.r. data for sugar-cane d-xylans

Arabir	noglucuron	oxylan	O- <i>Ac</i>	etylarabine	oglucuronoxylan
Chemi (p.p.n	ical shift 1.)a	Assignment ^b	Chemi (p.p.n	cal shift 1.)a	Assignment ^b
171.6	C=O	4-O-Methyl-α-D-GlcpA-	169.2	C=O	Ac and 4-O-methyl-α-D-GlcpA-
107.4	C-1	α-L-Araf-	168.9	C=O	Ac
103.1		β-D-Xylp-	106.9	C-1	α-tAraf-
102.2	C-1	4)-β-D-Xyl <i>p</i> -	103.1		β-D-Xylp-
	C-1	4)- β -D-Xyl p -2-substituted	101.6	C-1	4)-β-D-Xylp-
99.3	C-1	4- O -Methyl- α -D-Glc p A-		C-1	4)- β -D-Xylp-2-substituted
86.6	C-4	α-L-Araf-	99.3	C-1	4 - O -Methyl- α -D-Glc p A-
82.1	C-3	4)- β -D-Xylp-3-substituted	86.0	C-4	α-L-Araf-
80.7	C-2	α-L-Araf-	82.6	C-3	4)- β -D-Xylp-3-substituted
	C-4	4-O-Methyl-α-D-GlcpA-	81.2	C-2	α-L-Araf-
78.4	C-3	α-L-Araf-	81.1	C-4	4- O -Methyl- α -D-Glc p A-
77.2	C-2	4)- β -D-Xylp-2-substituted	77.7	C-3	α-L-Araf-
76.9	C-3	β -D-Xyl p -	76.7	C-2	4)-β-D-Xylp-2-substituted
76.3	C-4	4)-β-D-Xylp-		C-3	β -D-Xyl p -
74.4	C-3	4)-β-D-Xylp-	75.3	C-4	4)-β-D-Xylp-
73.9	C-2	β-D-Xylp-	73.9	C-3	4)-β-D-Xylp-
73.2	C-2	4)-β-D-Xylp-	73.3	C-2	β-D-Xylp-
72.5	C-3	4-O-Methyl- α -D-Glc p A-	72.5	C-2	4)-β-D-Xylp-
70,2	C-4	β-D-Xylp-	71.5	C-3	4- O -Methyl- α -D-Glc p A-
65.7	C-5	β -D-Xyl p -	70.2	C-4	B-D-Xylp-
63.6	C-5	4)-β-D-Xylp-	65.7	C-5	β-D-Xylp-
61.9	C-5	α-L-Araf-	63.1	C-5	4)-β-D-Xylp-
61.1	OCH_3	•	61.9	C-5	α-L-Araf-
	,		61.2	OCH_3	•
			20.8	CH ₃	of AcO-3-β-D-Xylp-
			20.1	CH_3	of AcO-2-β-D-Xylp-

^aDownfield from the signal for Me₄Si, measured against internal methanol (δ 50.15). ^bThe assignments were made by comparison with literature values (refs. 16–19) and data for 1–9.

The native O-acetylarabinoglucuronoxylan contained one acetyl group per 4.2 xylopyranosyl residues. Signals at 20.1 and 20.8 p.p.m. in the spectrum of P-1 and the ratio of their intensities suggest that the acetyl groups are located on O-3 and O-2 of the β -D-xylopyranosyl residues in the relative proportions 1.3:1.0. Similar patterns of substitution have been found in xylans from several tree species^{20,21}. Migration of acetyl groups during polysaccharide isolation cannot be excluded²².

The above results indicate sugar cane to contain water-insoluble substituted xylans in which there is a small variation in the proportions of the constituent sugars and in the pattern of substitution, although they have similar linkages.

Xylans from *Saccharum officinarum* have the same structural features as those reported for other Gramineae xylans. There are differences in chemical composition as well as size and shape of the macromolecules associated with side chains^{23–25}. Xylans present in sugar can exhibit heterogeneity as has been reported for primary cell-wall constituents such as xyloglucans^{26,27} and pectic polymers^{28,29}.

EXPERIMENTAL

General. — All evaporations were conducted under diminished pressure at <40°. Optical rotations (1-mL cell) were measured, unless stated otherwise, for solutions in aqueous 4% potassium hydroxide at $20 \pm 1^\circ$ with a Perkin–Elmer Model 141 polarimeter. G.l.c. was performed on a Hewlett–Packard Model 5711 A instrument with A, a column (305 × 0.3 cm) of 1% of XE-60 on Gas Chrom Z (80–100 mesh) at 130—150° at 1°/min, with N₂ as the carrier gas at 36 mL/min; B, a column (200 × 0.3 cm) of 3% of SP-2340 on Chromosorb WAW-DMCS (80–100 mesh) for 4 min at 180° then \rightarrow 210° at 2°/min, with N₂ at 30 mL/min; and C a column (400 × 0.45 cm) of 10% of Carbowax 400 on Chromosorb WAW (80–100 mesh) at 45°, with a N₂ flow rate of 17 mL/min. Column A was used for quantitative analysis of sugar trifluoroacetates³⁰.

G.l.c.-m.s. was carried out with a JGC-20 K gas chromatograph fitted with column B, and with helium (inlet pressure, 101.3 kPa) as the carrier gas. Mass spectra were obtained at 23 eV and an emission current of 300 μ A, using a JMS D (JEOL) spectrometer. The inlet temperature was 220° and that of the ionising chamber 200°.

Descending p.c. was conducted on Whatman No. 1 and 3MM papers with D, ethyl acetate-pyridine-water (8:2:1); E, ethyl acetate-acetic acid-water (18:7:8); F, 1-butanol-ethanol-water-ammonia (4:1:5:trace). Aniline hydrogen phthalate was used for detection. The mobilities ($R_{\rm Xyl}$ and $R_{\rm GlcA}$, respectively) are expressed relative to those of D-xylose and D-glucuronic acid. The retention times (T) of the methylated alditol acetates are given relative to that of 1,5-di-O-acetyl-2,3,4,6-tetra-O-methyl-D-glucitol. Uronic acids were determined by the modified carbazole method³¹ in the fraction eluted from Dowex 1 (AcO⁻) resin with 6M acetic acid after total hydrolysis, and acetyl contents by g.l.c. on column C.

 13 C-N.m.r. spectra were recorded for solutions in D₂O and (CD₃)₂SO with a Bruker AM-300 FT-spectrometer operating at 75.46 MHz in the noise-decoupled mode with deuterium lock. A spectral width of 17.241 Hz was used, 16k was used for the acquisition, and F.t. was with 32k (zero filling), an acquisition time of 0.48 s, and a relaxation delay of 3 s. Spectra were calibrated in D₂O (internal MeOH, vs. Me₄Si at 50.15 p.p.m.) and (CD₃)₂SO (signal of the solvent vs. Me₄Si at 39.4 p.p.m.).

Determination of molecular weight. — The number-average molecular weights (\overline{M}_n) were determined by osmotic pressure measurements on solutions in Me₂SO with a Knauer Membrane Osmometer and with Schleicher-Schüll membranes at 35°. The value for $(h/w)_{w=0}$ for AGX was 10.6 where h was the osmotic height in cm of the solution and w was the concentration in g/kg of the solution, giving an \overline{M}_n value of 22,300 (average d.p., 168). The \overline{M}_n values for AX and P-1 were 21,180 and 18,980, respectively.

Hydrolysis of the polysaccharides. — Complete hydrolysis of the polysaccharides was achieved by heating the sample (10 mg) with aqueous 90% formic acid (1 mL) for 8 h at 100°. After evaporation of the formic acid, the hydrolysate was

applied to a column (1 \times 10 cm) of Dowex 1 (AcO⁻) resin and eluted with water (200 mL) and then 6M acetic acid (100 mL). Each eluate was concentrated to dryness. The sample E and bagasse were treated with aqueous 72% sulfuric acid for 1 h at 20°, and the hydrolysate was then diluted to 3% acid, heated for 4 h at 100°, cooled, neutralised (BaCO₃), filtered, treated with Dowex 50 (H⁺) resin, and concentrated. The monosaccharides in the hydrolysates were converted into the corresponding alditol trifluoroacetates which were analysed by g.I.c. (column A).

The oligosaccharides (0.1-2 mg) were hydrolysed completely with aqueous 90% formic acid (0.1-0.5 mL) for 4 h at 100° . The products in the hydrolysates were examined by p.c. (solvent D and E) and g.l.c. of the derived alditol acetates.

The acidic oligosaccharides were treated with methanolic 3% HCl (12 h, 90°), and the products were reduced (NaBH₄ in dry methanol, 16 h), then hydrolysed with aqueous 90% formic acid (4 h, 100°), converted into the corresponding alditol trifluoroacetates, and analysed by g.l.c.

Plant material. — Sugar cane (Saccharum officinarum, var. J-60-5; 23.2 kg) cultivated in the Sugar Cane Experimental Centre, Bauta, Havana, was cut and the juice was separated by rolling and then extraction with water (twice, 24 h, 40°). The fibrous residue was defibrated in a Sproud Waldrom disk refinator to yield 3.5 kg of air-dried cell-wall material (CWM). Standard analytical methods showed that the CWM contained (%) cellulose (44.9), lignin (Klason; 20.5), uronic anhydride (1.0), acetyl (0.3), galactose (0.4), glucose (50.4), mannose (1.7), arabinose (3.3), and xylose (23.2).

Fractionation of CWM. — Sawdust (40-60 mesh) obtained from freshly cut sugar-cane stalks was extracted with benzene-ethanol (35:65) for 4 h, then stirred (30 mL/g) with (a) water [4 h, 80° (bath)], and (b) aqueous 0.5% ammonium oxalate (4 h, 80°) to remove pectic substances. The depectinated residue (60 g) was treated with acidified sodium chlorite solution⁴ for 2 h at 25°, and the resulting holocellulose (49.02 g, 81.7% of the CWM) was extracted as follows. (a) The holocellulose (20.0 g) was extracted (4 h, 25°, under N₂) successively with aqueous 15% potassium hydroxide and then aqueous 17.5% sodium hydroxide containing 4% of boric acid. Each extract was filtered and neutralised with acetic acid to give the insoluble fractions C-1 and D-1, respectively, which were collected by centrifugation at 1.8×10^4 r.p.m. The soluble fractions, C-2 and D-2, were obtained by precipitation with ethanol (10 vol.) following exhaustive dialysis against distilled water. (b) Holocellulose (28.0 g) was extracted (30 mL/g) twice with methyl sulfoxide⁶ (72 h, 25°) to give polysaccharide P-1 (1.13 g, 3.3% of the CWM), $[\alpha]_D$ -100° (c 1) (Found: uronic acid, 0.7; acetyl, 6.8%; $\overline{M}_{\rm n}$ 18,980), which, on hydrolysis, gave xylose, arabinose, and glucose in the molar ratios 10.0:0.7:0.1. The Me₂SO-insoluble residue (26.8 g) was further extracted with water (30 mL/g, 72 h, 25°) to yield an insoluble residue (26.3 g) and soluble material which was dialysed exhaustively against distilled water and then freeze-dried to yield polysaccharide P-2 (377 mg, 1.1% of the CWM), $[\alpha]_D$ -79° (c 1) (Found: uronic acid, traces; acetyl, 3.9%), which, on hydrolysis, gave xylose, arabinose, glucose, galactose,

and mannose in the molar ratios 10.0:1.5:0.5:0.4:0.1. Data on these fractions are given in Table II.

Arabinoglucuronoxylan. — (a) Isolation. Fraction C-I (1 g), $[\alpha]_D$ = -136° (c 1), was fractionated by precipitation from solution in aqueous 4% potassium hydroxide (100 mL) by addition of Fehling's solution⁵ (10–15 mL). The resulting precipitate was dispersed in water (50 mL), and 0.2m hydrochloric acid was added, followed by precipitation with ethanol (5 vol.). Repetition of this procedure twice gave no further change in the sugar composition of the arabinoglucuronoxylan AGX (807 mg, 24.7% of the CWM), $[\alpha]_D$ = -143° (c 1), which, on hydrolysis, gave xylose and arabinose in the molar ratio 10.0:0.6 (Found: uronic acid, 1.0%; \overline{M}_n 22,300).

Fraction *D-1* (200 mg), $[\alpha]_D = 111^\circ$ (*c* 1), was treated, as described above for *C-1*, to give arabinoxylan *AX* (169.7 mg, 0.7% of the CWM), $[\alpha]_D = 132^\circ$ (*c* 1), which, on hydrolysis, gave xylose and arabinose in the molar ratio 10.0:0.5 (Found: uronic acid, 0%; \overline{M}_n 21,180).

(b) Partial hydrolysis. AGX (2.0 g) was treated twice with 0.25M trifluoroacetic acid (60 mL) for 2 h at 100°. The hydrolysate was concentrated after each treatment and the part to be hydrolysed further was collected by centrifugation. The combined hydrolysates were concentrated and the residue was resolved into neutral and acidic fractions on a column of Dowex 1 (AcO⁻) resin.

The low-molecular-weight neutral fragments (684 mg), after fractionation on a column (2.5 \times 100 cm) of Sephadex G-10 with water, gave xylose, arabinose, and a mixture (174.9 mg) of oligosaccharides. Preparative p.c. on Whatman No. 3MM paper (solvent D) gave 1–5 (Tables III and IV). The acidic fragments (82 mg) were subjected to preparative p.c. (solvent E) to give 6–9 (Table V). ¹³C-N.m.r. data were recorded for each oligosaccharide. The neutral oligosaccharides (4 mg) were each methylated with methyl iodide (2 mL) and sodium hydride (20 mg) in N,N-dimethylformamide (2 mL). The products were hydrolysed with 0.1M hydrochloric acid, the hydrolysate was neutralised, and the products were analysed as alditol acetates by g.l.c. ¹⁴ (column B).

Oligosaccharides **6–9** were also methylated (Hakomori), and the products were reduced (NaBH₄), hydrolysed, and analysed by g.l.c. as the partially methylated alditol acetates.

(c) Methylation analysis. A portion (55 mg) of AGX was methylated once by the Hakomori method¹², then twice by the Purdie method¹³, to give a product (53 mg), $[\alpha]_D$ -74° (c 1, chloroform), showing no i.r. absorption for hydroxyl (Found: methoxyl, 37.6%; \overline{P}_n 76). Likewise, AX (30 mg) was methylalted to give a product (25 mg) having $[\alpha]_D$ -69° (c 0.5, chloroform) (Found: methoxyl, 38.1%; \overline{P}_n 84).

The methylated polysaccharides (5 mg of each) were treated with aqueous 90% formic acid (2 mL) for 3 h at 100°, the hydrolysates were concentrated to dryness, and each residue was hydrolysed with 2M hydrochloric acid (2 mL) for 6 h at 100°. P.c. then revealed mono-, di-, and tri-O-methyl saccharides with $R_{\rm Xyl}$ 1.00, 0.94, 0.89, 0.82, and 0.52 (solvent F), which were converted into their alditol acetates. Analysis by g.l.c.-m.s. ¹⁴ gave the results in Table VI.

- O-Acetylarabinoglucuronoxylan. (a) Methylation. P-I (10 mg), $[\alpha]_D 100^\circ$ (c 1), was methylated (Hakomori) to give a product (8 mg) having $[\alpha]_D 78^\circ$ (c 1, chloroform) (Found: OMe, 37.8%) (see Table VI, column 3).
- (b) Location of the O-acetyl groups. The location of the acetyl groups in P-1 was determined qualitatively by 13 C-n.m.r. spectroscopy, employing the data for methyl 2-O- and 3-O-acetyl- β -D-xylopyranosides 19 .

ACKNOWLEDGMENT

We thank Mr. G. Košický for the molecular weight determinations and the optical rotation measurements.

REFERENCES

- 1 S. R. PATHAK, Proc. Int. Soc. Sugar Cane Technologist, 11 (1962) 1211-1216.
- 2 S. R. PATHAK AND V. R. SRINIVASAN, Indian J. Chem., 2 (1964) 365-367.
- 3 J. D. BLAKE, M. L. CLARKE, AND P.-E. JANSSON, Carbohydr. Res., 115 (1983) 265-272.
- 4 F. SAAVEDRA AND D. CORDERO, Sobre los Derivados ICIDCA, 16 (1982) 46-51.
- 5 J. K. N. JONES AND R. J. STOODLEY, Methods Carbohydr. Chem., 5 (1965) 36-38.
- 6 E. HÄGGLUND, B. LINDBERG, AND J. MCPHERSON, Acta Chem. Scand., 10 (1956) 1160-1164.
- 7 R. L. WHISTLER, H. E. CONRAD, AND L. HOUGH, J. Am. Chem. Soc., 76 (1954) 1668-1670.
- 8 G. O. ASPINALL AND I. M. CAIRNCROSS, J. Chem. Soc., (1960) 3877-3881.
- 9 H. C. SRIVASTAVA, C. T. BISHOP, AND G. A. ADAMS, J. Org. Chem., 26 (1961) 3958-3960.
- 10 G. O. ASPINALL AND E. G. MEEK, J. Chem. Soc., (1956) 3830–3834.
- 11 G. O. ASPINALL AND K. C. B. WILKIE, J. Chem. Soc., (1956) 1072-1076.
- 12 S. HAKOMORI, J. Biochem. (Tokyo), 55 (1964) 205-208.
- 13 E. L. HIRST AND E. PERCIVAL, Methods Carbohydr. Chem., 5 (1965) 287-296.
- 14 P.-E. Jansson, L. Kenne, H. Liedgren, B. Lindberg, and J. Lönngren, Chem. Commun. Univ. Stockholm, 8 (1976) 1–75.
- 15 P. A. J. GORIN AND M. MAZUREK, Can. J. Chem., 53 (1975) 1212-1223.
- 16 P. Kováč, J. Hirsch, A. S. Shashkov, A. I. Usov, and S. Yarotsky, *Carbohydr. Res.*, 85 (1980) 177–185.
- 17 J. HIRSCH, E. PETRÁKOVÁ, AND J. SCHRAML, Carbohydr. Res., 131 (1984) 219-226.
- 18 J. P. UTILLE, P. KOVÁČ, F. SAURIOL, AND A. S. PERLIN, Carbohydr. Res., 154 (1986) 251–258.
- 19 Š. KARÁCSONYI, J. ALFÖLDI, M. KUBAČKOVÁ, AND Ľ. STUPKA, Cellul. Chem. Technol., 17 (1983) 637–645.
- 20 T. E. TIMELL, Adv. Carbohydr. Chem., 19 (1964) 247-302.
- 21 B. LINDBERG, K.-G. ROSELL, AND S. SVENSSON, Sven. Papperstidn., 76 (1973) 30-32.
- 22 B. LINDBERG AND K. C. ROSELL, Sven. Papperstidn., 77 (1974) 286-287.
- 23 G. O. ASPINALL, Adv. Carbohydr. Chem., 14 (1959) 429-468.
- 24 K. C. B. WILKIE, Adv. Carbohydr. Chem. Biochem., 36 (1979) 215-264.
- 25 A. M. STEPHEN, in G. O. ASPINALL (Ed.), The Polysaccharides, Vol. 2, Academic Press, New York, 1983, pp. 98–193.
- 26 J.-P. Joseleau and G. Chambat, Plant Physiol., 74 (1984) 694-700.
- 27 P. RUPEREZ, R. R. SELVENDRAN, AND B. J. H. STEVENS, Carbohydr. Res., 142 (1985) 107-113.
- 28 A. G. DARVILL, M. MCNEIL, AND P. ALBERSHEIM, Plant Physiol., 62 (1978) 418-422.
- 29 M. McNeil, A. G. Darvill, and P. Albersheim, Plant Physiol., 66 (1980) 1128-1134.
- 30 Y. Shapira, Nature (London), 222 (1969) 792-793.
- 31 T. BITTER AND H. M. MUIR, Anal. Biochem., 4 (1962) 330-334.